## MATH3104 Lecture 1 (Bracken)

-Diffusion: random movement of molecules or small particles in a gas or liquid, due to thermal energy of surrounding molecules.
-Recall: Absolute (or Kelvin) temperature scale.

$$
27^{\circ} \mathrm{C} \approx 300^{\circ} \mathrm{K}
$$

- Particle in fluid that is in thermal equilibrium at $T^{0} K$, has average K.E. associated with motion along each coord. axis equal to $\frac{1}{2} k T$
- so $\frac{3}{2} k T$ in total.
- Here $k$ is Boltzmann's constant

$$
k \approx 1.38 \times 10^{-16} \quad \mathrm{gm}(\mathrm{~cm} / \mathrm{sec})^{2} /{ }^{\circ} \mathrm{K}
$$

- When $T=300^{\circ} K, \quad k T \approx 4.14 \times 10^{-14} \mathrm{gm}(\mathrm{cm} / \mathrm{sec})^{2} \quad(\mathrm{ergs})$.
- a very small energy on human scales: a 70 kg human walking at 5 kph has a K.E.

$$
\frac{1}{2} m v^{2} \approx 7 \times 10^{8} \mathrm{ergs}
$$

- However, consider a molecule of the enzyme/protein lysozyme, in water at $300^{\circ} \mathrm{K}$. (Lysozyme is found in egg-white, tears, ...)
-Mass? Molecular weight $\approx 14,000 \mathrm{gm}$
$=$ mass of $N_{A}$ molecules, where $N_{A}=\underline{\text { Avogadro Number } \approx 6 \times 10^{23}}$

So now

$$
m \approx \frac{14000}{6 \times 10^{23}} \approx 2.3 \times 10^{-20} \mathrm{gm}
$$

-Speed? $\quad \frac{1}{2} m v_{x}^{2} \approx \frac{1}{2} k T$

$$
\Rightarrow v_{x} \approx \sqrt{\frac{k T}{m}} \approx \sqrt{\frac{4.14 \times 10^{-14}}{2.3 \times 10^{-20}}} \approx 1.3 \times 10^{3} \mathrm{~cm} / \mathrm{sec} \quad(\approx 45 \mathrm{kph})
$$

- would cross swimming pool in about 1 sec.
- Each water molecule has a similar average K.E. along each coordinate axis. But now
$M W=18 \mathrm{gm}$, so $m \approx(18) /\left(6 \times 10^{23}\right) \approx 3 \times 10^{-23} \mathrm{gm}$
$\Rightarrow v_{x}$ about $\sqrt{\frac{14000}{18}} \approx 30$ times greater.
- would cross pool in about $1 / 30$ sec.
- Of course, this is not what happens. Molecules collide repeatedly and get redirected. The lysozyme molecule in water is forced to conduct a random walk.
- A small cloud of such particles will wander about and spread - this is diffusion.
-Let's consider a simple model of this process:


## The one-dimensional random walk

A particle ('the walker') starts at $x=0$ at time $t=0$. After each interval of time $\tau$, it receives a kick and moves one step of length $\delta$ to L or R along the $X$-axis, each with probability $1 / 2$.
(Toss a coin at each stage!)
e.g. after 3 steps (at $t=3 \tau$ ), particle could be at


- See the pattern:-

$\frac{1}{2}$ - $\frac{1}{2}$ -

$$
\frac{1}{8} \cdot \frac{3}{8}
$$

$$
\frac{1}{8}
$$

$$
\begin{aligned}
& t=0 \\
& t=\tau \\
& t=2 \tau \\
& t=3 \tau \\
& t=4 \tau
\end{aligned}
$$

$$
\frac{1}{4} \cdot \frac{1}{2} \cdot \frac{1}{4} \cdot \quad t=2 \tau
$$

Pascal Triangle

$$
\frac{3}{8}=\frac{1}{2}\left(\frac{1}{2}+\frac{1}{4}\right) \text { etc. }
$$

After $N$ steps ( $t=N \tau$ ), particle could be at any of

$$
x=N \delta, \quad x=(N-2) \delta, \quad \ldots \quad x=-(N-2) \delta, \quad x=-N \delta
$$

i.e. particle is at

$$
x=m \delta, \quad m \in\{N, N-2, \cdots-(N-2),-N\}
$$

For a given $m$, this requires a sequence of steps of which $r$ are to the RIGHT and $l$ are to the LEFT, with

$$
r-l=m .
$$

Since

$$
r+l=N
$$

it follows that

$$
r=\frac{N+m}{2}, \quad l=\frac{N-m}{2} .
$$

-The probability of any one sequence of $N$ steps is $(1 / 2)^{N}$.

So, probability that $x=m \delta$ after $N$ steps is
$P(m, N)=\left(\frac{1}{2}\right)^{N}$ [ No. of sequences of length $N$ with $r=(N+m) / 2$ ]

$$
\begin{aligned}
& =\left(\frac{1}{2}\right)^{N} \text { [ No. of ways of getting } r \text { Heads in } N \text { coin tosses] } \\
& =\left(\frac{1}{2}\right)^{N} C_{r}^{N},
\end{aligned}
$$

where $C_{r}^{N}=\frac{N!}{r!(N-r)!} \quad(=N \quad$ choose $r)$.

So we have

$$
P(m, N)=\left(\frac{1}{2}\right)^{N} \frac{N!}{r!(N-r)!}=\left(\frac{1}{2}\right)^{N} \frac{N!}{\left(\frac{N+m}{2}\right)!\left(\frac{N-m}{2}\right)!}
$$

$$
\underline{\underline{\text { EX: }} \quad} \quad N=3, \quad m=-1 \quad(\Rightarrow r=1, \quad l=2)
$$

$$
P(-1,3)=\left(\frac{1}{2}\right)^{3} \frac{3!}{1!2!}=\left(\frac{1}{2}\right)^{3} 3=\frac{3}{8}
$$

— as on page 1.6
************************** * * * * * *

Note that we must have

$$
\sum_{m=-N}^{N} \prime^{\prime} P(m, N)=\sum_{m=-N}^{N} \prime\left(\frac{1}{2}\right)^{N} \frac{N!}{\left(\frac{N+m}{2}\right)!\left(\frac{N-m}{2}\right)!}=1
$$

as in the example on p. 1.6. Can you see how to prove it in the general case? (Binomial Theorem!)
(Here $\sum^{\prime}$ means sum over $\left.m=-N,-(N-2), \ldots, N-2, N.\right)$
-Where is the particle on average after $N$ steps?

$$
\langle x\rangle=\sum_{m=-N}^{N}{ }^{\prime} P(m, N) m \delta=0
$$

because $P(-m, N)=P(m, N)$
— particle is just as likely to go $L$ or $R$ at each step - on average it gets nowhere!

More interesting is the mean-square displacement of the particle from its mean position:-

$$
\begin{aligned}
& \left\langle(x-\langle x\rangle)^{2}\right\rangle=\left\langle x^{2}-2\langle x\rangle x+\langle x\rangle^{2}\right\rangle \\
& =\left\langle x^{2}\right\rangle-2\langle x\rangle\langle x\rangle+\langle x\rangle^{2} \\
& =\left\langle x^{2}\right\rangle-\langle x\rangle^{2} . \text { This reduces to }\left\langle x^{2}\right\rangle \text { here, because }\langle x\rangle=0 .
\end{aligned}
$$

We have

$$
\begin{aligned}
& \left\langle x^{2}\right\rangle \\
& \begin{array}{cc}
(1)(0)^{2}=0 & \text { No. of steps } \\
\begin{array}{cc}
\left(\frac{1}{2}\right)(-\delta)^{2}+\left(\frac{1}{2}\right)(\delta)^{2}=\delta^{2} & 0 \\
\hline\left(\frac{1}{4}\right)(-2 \delta)^{2}+\left(\frac{1}{2}\right)(0)^{2}+\frac{1}{4}(2 \delta)^{2} & 1 \\
= & 2 \delta^{2}
\end{array} \\
\begin{array}{rl}
\left(\frac{1}{8}\right)(-3 \delta)^{2}+\left(\frac{3}{8}\right)(-\delta)^{2}+\left(\frac{3}{8}\right)(\delta)^{2}+\left(\frac{1}{8}\right)(3 \delta)^{2} & 2 \\
= & 3 \delta^{2}
\end{array}
\end{array}
\end{aligned}
$$

The pattern is clear: after $N$ steps,

$$
\left\langle x^{2}\right\rangle=N \delta^{2}
$$

ie. $\sum_{m=-N}^{N} \prime P(m, N)(m \delta)^{2}=N \delta^{2}$ - but can you prove it?
-The root-mean-square displacement $\sqrt{\left\langle(x-\langle x\rangle)^{2}\right\rangle}$ is a convenient measure of how far we expect the particle to be from its mean position.

We have $\sqrt{\left\langle(x-\langle x\rangle)^{2}\right\rangle}=\sqrt{\left\langle x^{2}\right\rangle}=\sqrt{N} \delta$ after $N$ steps.

This is an important and characteristic feature of the random walk!

After 100 steps, each of length $\delta$, we expect the particle to be about $10 \delta$ from its starting point.
After 10, 000 steps, we expect it to be about $100 \delta$ away, and so on.

Note that we are talking about average behaviour. No two realisations of a random walk will look the same in general: see the following figures:-


-Consider again

$$
P(m, N)=\left(\frac{1}{2}\right)^{N} \frac{N!}{\left(\frac{N+m}{2}\right)!\left(\frac{N-m}{2}\right)!} .
$$

Using Stirling's approximation for $K$ ! when $K$ is large, we can show that when $N$ and $|m|$ are large, with $m^{2} / N$ not too large, then

$$
P(m, N) \approx \sqrt{\frac{2}{\pi N}} e^{-m^{2} / 2 N}
$$

This is a good approximation even for quite small values of $|m|$ and $N$ :-

$$
\begin{array}{cccc}
(m, N) & \begin{array}{lll}
P(m, N) & \begin{array}{c}
\text { Relative } \\
\text { Error }
\end{array} \\
( \pm 2,10) & 0.2051 & 0.2066
\end{array} & 0.73 \% \\
( \pm 5,27) & 0.09714 & 0.09668 & 0.5 \% \\
( \pm 200,10000) & 0.0010799 & 0.0010798 & 0.01 \%
\end{array}
$$



## 31 realizations


-What about random walks in two and three dimensions?

Again we find that the mean displacement after $N$ steps is zero, and the rooot-mean-square displacement is proportional to $N$.

See the next two figures, showing realisations of a 2-D random walk with $N=100$ and $N=1000$, respectively.
[We have assumed that after each time interval of length $\tau$, the particle steps a distance $\delta$ in an arbitrary direction, making an angle with the $X$-axis that is uniformly distributed over $[0,2 \pi)$.]



Reading: H.C. Berg, Random Walks in Biology, Chapter 1.

