•<u>Diffusion</u>: random movement of molecules or small particles in a gas or liquid, due to thermal energy of surrounding molecules.

•Recall: $\underline{Absolute}$ (or \underline{Kelvin}) temperature scale.

 $27^{\circ}C \approx 300^{\circ}K$

• Particle in fluid that is in thermal equilibrium at $T^o K$, has <u>average</u> K.E. associated with motion along each coord. axis equal to $\frac{1}{2}kT$

— so $\frac{3}{2}kT$ in total.

• Here k is Boltzmann's constant

 $k \approx 1.38 \times 10^{-16} \quad gm(cm/sec)^2/{^oK}$

- When $T = 300^{\circ}K$, $kT \approx 4.14 \times 10^{-14} gm (cm/sec)^2$ (ergs).
- a very small energy on human scales: a 70kg human walking at 5kph has a K.E.

$$\frac{1}{2}mv^2 \approx 7 \times 10^8 ergs$$

•However, consider a molecule of the enzyme/protein $\underline{lysozyme}$, in water at $300^{\circ}K$. (Lysozyme is found in egg-white, tears, ...)

• Mass? Molecular weight $\approx 14,000 gm$

= mass of N_A molecules, where $N_A = \mathbf{Avogadro Number} \approx 6 \times 10^{23}$

So now

$$m \approx \frac{14000}{6 \times 10^{23}} \approx 2.3 \times 10^{-20} gm$$

•**Speed?** $\frac{1}{2}mv_x^2 \approx \frac{1}{2}kT$

$$\Rightarrow v_x \approx \sqrt{\frac{kT}{m}} \approx \sqrt{\frac{4.14 \times 10^{-14}}{2.3 \times 10^{-20}}} \approx 1.3 \times 10^3 cm/sec \quad (\approx 45 kph)$$

— would cross swimming pool in about 1sec.

•Each water molecule has a similar <u>average</u> K.E. along each coordinate axis. But now

MW=18gm , so $m\approx(18)/(6\times10^{23})\approx3\times10^{-23}gm$

 $\Rightarrow v_x$ about $\sqrt{\frac{14000}{18}} \approx 30$ times greater.

— would cross pool in about 1/30sec.

•Of course, this is not what happens. Molecules collide repeatedly and get redirected. The lysozyme molecule in water is forced to conduct a <u>random walk</u>.

•A small cloud of such particles will wander about and spread — this is <u>diffusion</u>.

•Let's consider a simple model of this process:

The one-dimensional random walk

A particle ('the walker') starts at x = 0 at time t = 0. After each interval of time τ , it receives a kick and moves one step of length δ to L or R along the X-axis, each with probability 1/2.

(Toss a coin at each stage!)

e.g. after 3 steps (at t=32), particle could be at Probability Steps $\left(\frac{z}{1}\right)_3 = \frac{2}{12}$ x= 38 RRR SRRL RLR LRR x = 8 $(\frac{1}{2})^{3} = \frac{3}{2}$ LRL $(\frac{1}{2})^{3} = \frac{1}{8}$ X= - 8 RLL $(\frac{1}{2})^{3} = \frac{1}{2}$ X = - 38 LLL sum = - particle must be somewhere!

. See the pattern : --98-35-28-8 8 28 38 48 × t=o t=T t=22 t = 3T4. $t=4\tau$ Pascal Triangle 3 = ± (±+ +) etc.

After N steps $(t = N\tau)$, particle could be at any of

$$x = N\delta$$
, $x = (N - 2)\delta$, ... $x = -(N - 2)\delta$, $x = -N\delta$

i.e. particle is at

$$x = m\delta$$
, $m \in \{N, N - 2, \dots - (N - 2), -N\}$

For a given m, this requires a sequence of steps of which r are to the RIGHT and l are to the LEFT, with

$$r-l=m$$
 .

Since

$$r+l=N\,,$$

it follows that

$$r = \frac{N+m}{2}, \qquad l = \frac{N-m}{2}.$$

• The probability of any <u>one</u> sequence of N steps is $(1/2)^N$.

So, probability that $x = m\delta$ after N steps is

 $P(m,N) = \left(\frac{1}{2}\right)^N$ [No. of sequences of length N with r = (N+m)/2] $= \left(\frac{1}{2}\right)^N$ [No. of ways of getting r Heads in N coin tosses] $= \left(\frac{1}{2}\right)^{N} C_{r}^{N}$ where $C_r^N = \frac{N!}{r!(N-r)!}$ (= N choose r). So we have $P(m,N) = \left(\frac{1}{2}\right)^N \frac{N!}{r!(N-r)!} = \left(\frac{1}{2}\right)^N \frac{N!}{(\frac{N+m}{2})!(\frac{N-m}{2})!}.$

EX:
$$N = 3$$
, $m = -1$ $(\Rightarrow r = 1, l = 2)$

$$P(-1,3) = \left(\frac{1}{2}\right)^3 \frac{3!}{1!\,2!} = \left(\frac{1}{2}\right)^3 \,3 = \frac{3}{8}$$

— as on page 1.6

Note that we must have

$$\sum_{m=-N}^{N} {'P(m,N)} = \sum_{m=-N}^{N} {'\left(\frac{1}{2}\right)^{N} \frac{N!}{(\frac{N+m}{2})!(\frac{N-m}{2})!}} = 1,$$

as in the example on p. 1.6. Can you see how to prove it in the general case? (Binomial Theorem!)

(Here \sum' means sum over $m = -N, -(N-2), \ldots, N-2, N$.)

•Where is the particle on average after N steps?

$$\langle x \rangle = \sum_{m=-N}^{N} {}' P(m,N) m \delta = 0$$

because P(-m, N) = P(m, N)

— particle is just as likely to go L or R at each step — $\underline{\text{on average}}$ it gets nowhere!

More interesting is the <u>mean-square displacement</u> of the particle from its mean position:-

$$\begin{split} &\langle (x - \langle x \rangle)^2 \rangle = \langle x^2 - 2 \langle x \rangle x + \langle x \rangle^2 \rangle \\ &= \langle x^2 \rangle - 2 \langle x \rangle \langle x \rangle + \langle x \rangle^2 \\ &= \langle x^2 \rangle - \langle x \rangle^2. \text{ This reduces to } \langle x^2 \rangle \text{ here, because } \langle x \rangle = 0. \end{split}$$

We have

< x ² >	No. of steps
$(1)(0)^{2} = 0$	0
$(\frac{1}{2})(-\delta)^{2} + (\frac{1}{2})(\delta)^{2} = \delta^{2}$	1
$(\frac{1}{4})(-2\delta)^{2} + (\frac{1}{4})(0)^{2} + \frac{1}{4}(2\delta)^{2}$ = $2\delta^{2}$	5) ² 2
$(\frac{1}{8})(-3\delta)^{2} + (\frac{3}{8})(-\delta)^{2} + (\frac{3}{8})(-\delta)^{2}$ = $3\delta^{2}$	$(8)^{2} + (\frac{1}{8})(38)^{2}$ 3
The pattern is $\langle x^{*} \rangle = 1$	clear: after N steps N 8 ²
ie. Zi 'P(m, r me-N	$N)(mS)^{2} = NS^{2}$
- bu	t can you prove it?

•The root-mean-square displacement $\sqrt{\langle (x - \langle x \rangle)^2 \rangle}$ is a convenient measure of <u>how far</u> we expect the particle to be <u>from its mean position</u>.

We have $\sqrt{\langle (x - \langle x \rangle)^2 \rangle} = \sqrt{\langle x^2 \rangle} = \sqrt{N} \,\delta$ after N steps.

This is an important and characteristic feature of the random walk!

After 100 steps, each of length δ , we expect the particle to be <u>about</u> 10δ from its starting point.

After 10,000 steps, we expect it to be <u>about</u> 100δ away, and so on.

Note that we are talking about <u>average behaviour</u>. No two <u>realisations</u> of a random walk will look the same in general: see the following figures:-





•Consider again

$$P(m, N) = \left(\frac{1}{2}\right)^{N} \frac{N!}{(\frac{N+m}{2})!(\frac{N-m}{2})!}.$$

Using Stirling's approximation for K! when K is large, we can show that when N and |m| are large, with m^2/N not too large, then

$$P(m,N) \approx \sqrt{\frac{2}{\pi N}} e^{-m^2/2N}$$
.

This is a good approximation even for quite small values of |m| and N:-

(m, N)	P(m,N)	$\int \frac{2}{\pi N} e^{-m^2/2N}$	Relative Error
(±2,10)	0.2051	0.2066	0.73%

(±5,27) 0.09714 0.09668 0.5%

(±200, 10000) 0.0010799 0.0010798 0.01%





•What about random walks in two and three dimensions?

Again we find that the mean displacement after N steps is zero, and the rooot-mean-square displacement is proportional to N.

See the next two figures, showing realisations of a 2-D random walk with N = 100 and N = 1000, respectively.

[We have assumed that after each time interval of length τ , the particle steps a distance δ in an arbitrary direction, making an angle with the X-axis that is uniformly distributed over $[0, 2\pi)$.]





Reading: H.C. Berg, Random Walks in Biology, Chapter 1.