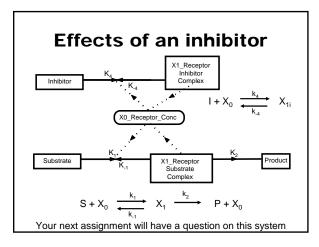
### 📽 МАТН1070

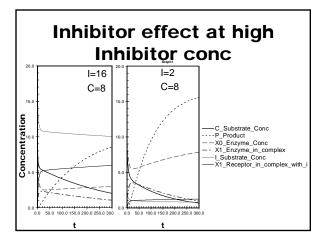
L5 Inhibitor Interactions

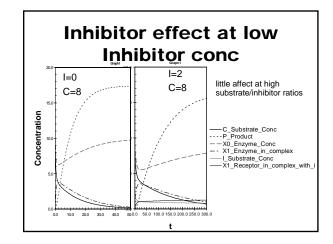
#### Euler's Method for solving ODEs This is the simplest approximation method. http://www.physics.uq.edu.au/people/jones/ph362/cphys/node1.html For $\frac{dy}{dt} = f(y);$ $y_{n+1} = y_n + hf(y_n)$ , where h is the step size. Take the function $\frac{dy}{dt} = ky_{dt}$ Its approximation is: $\frac{Y_{h+1}Y_{h} = kY_{h}}{dt}_{t}$ $y_{h+1}Y_{h} = dt * kY_{h}$ $y_{h+1} = y_{h} + dt * kY_{h}$

# Effects of an inhibitor on enzyme kinetics

- An inhibitor may competitively bind with an enzyme required for a substrate.
- Under high concentrations of inhibitor and low concentrations of substrate (and limiting levels of enzyme) the conversion of substrate to product is very low.
- Competitive inhibition can be overcome at a sufficiently high substrate concentration.







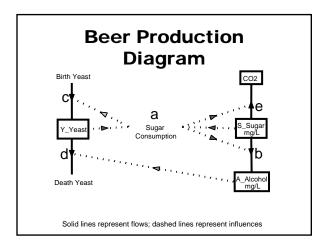
#### **Beer Production**

- An example that includes saturation and mass action.
- Simplification: brewing beer involves putting yeast and sugar in a vessel so that alcohol is produced as a by poduct of the metabolism of yeast.
- FACTS:
  - There is a finite amount of sugar at the start.
  - Sugar enhances yeast formation through mass action with yeast but this causes sugar breakdown into fractions of alcohol and CO<sub>2</sub>.
  - Excessive alcohol will kill yeast cells.

#### **Beer Production**

Assumptions:

- the rates of sugar consumption, CO<sub>2</sub> and alcohol production, and yeast mortality due to alcohol follow mass action laws.
- rate of alcohol production is proportional to the rate of sugar consumption.



## What parameters are required?

- a. Rate of sugar uptake & consumption by yeast
- b. Rate of alcohol produced from sugar consumption by yeast
- c. Rate of yeast cell formation per unit of sugar consumed
- d. Death rate of yeast cells per unit of alcohol
- e. Fraction of sugar breakdown that yields CO<sub>2</sub>

